[2,3] Sigmatropic Rearrangement of 1-Vinylic Tetrahydroisoquinoline N-Ylides and N-Oxides.

T. Samuel Baileva, John B. Bremnerb* and John A. Carverb

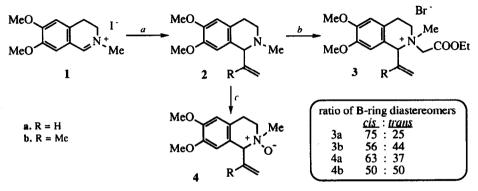
a. Department of Chemistry, University of Tasmania, GPO Box 252C, Hobart, Australia 7001.

b. Department of Chemistry, University of Wollongong, Northfields Ave., Wollongong, Australia 2522.

Abstract: [2.3] Signatropic rearrangement of the N-ylides from the 1-vinyl and 1-(1-propenyl) tetrahydroisoquinolinium salts **3a** and **3b** at room temperature in acetonitrile gave high yields of the functionalised 3-benzazonine derivatives **6a** (90%: E/Z = 19) and **6b** (93%: E/Z = 2) respectively. The N-oxide analogues **4a** and **4b** gave the 2.3-benzoxazepines **8a** and **8b**, together with the first representative of the 4.3-benzoxazonine system. **9b**, from **4b**.

2-Vinylpyrrolidine^{1a} and piperidine^{1b} N-ylides are known to undergo [2,3] sigmatropic rearrangements with ring expansion^{1c}. A [2,3] sigmatropic rearrangement was postulated as the initial step in the overall Stevens rearrangement of 1-phenyl tetrahydroisoquinoline N-ylide derivatives^{1d}. In contrast 2-vinylpiperidine N-oxide^{2a} underwent only the Meisenheimer rearrangement. More complex allylic N-oxide systems have, however, been reported to produce [2,3] rearrangement and ring expansion^{2b.c.d}. We now wish to report the analogous rearrangements of 1-vinylic tetrahydroisoquinoline N-ylide 5 and N-oxide 4 derivatives. The rearrangements provide both a new and convenient route to functionalised 3-benzazonines³ and also access to the previously undescribed 4,3-benzoxazonine system.

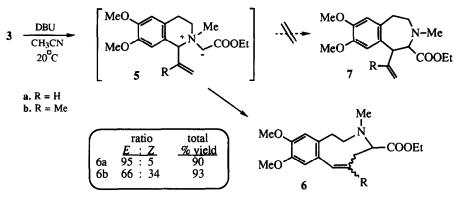
Scheme 1.



Reagents: a) BrMgC(R)CH2 b) BrCH2COOEt c) m-CPBA

The precursors required for the rearrangements were prepared from the iminium salt 1⁴. Addition of 1 to the appropriate alkenyl Grignard reagent in THF at -78°C, then stirring at RT for 15 hrs, gave the bases 2 in excellent yields (Scheme 1)⁵. The quaternary salts 3 were obtained by N-alkylation with neat ethyl bromoacetate, trituration of the crude product with ether and crystallisation from ethanol/THF. The N-oxides 4 were obtained by oxidation of 2 with m-CPBA in CH₂Cl₂ at 0-25°C and were utilised as crude oils. In both cases the products were observed by NMR to be mixtures of B ring diastereomers. By analogy with the piperidine system the major product was assumed to be the *cis* diastereomer, formed by axial attack on the nitrogen with the alkenyl substituent in the equatorial position.

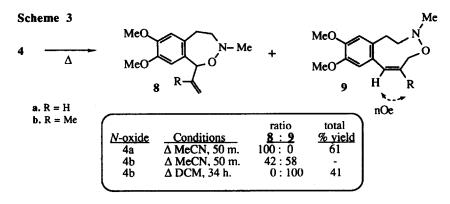
Scheme 2



The ylide rearrangements were found to produce only the 3-benzazonines 6 via the [2,3] sigmatropic mechanism, with none of the alternative Stevens rearrangement products 7 observed (Scheme 2). The ylides were generated by deprotonation of 3 with DBU. A rearranged fraction, as indicated by ¹H and ¹³C NMR, was obtained by chromatography on alumina with DCM/20% L.P.(60-80°). The diastereomeric products were then separated by reverse phase HPLC on a C18 column with MeCN/20% H₂O. Attempts to purify the mixture by PLC on silica resulted in loss of the *E*-benzazonines. The unsaturated 3-benzazonines were characterised⁶ by olefinic ¹³CH peaks at 128-135 δ , H7 at 6.3-6.7 δ and H4 at 3.4-3.6 δ . The observed selectivity towards the *E* isomer is consistent with the behaviour of the piperidine *N*-ylide system^{1b} and proposed [2,3] rearragement transition states^{1c}.

In contrast, the N-oxides 4 upon thermolysis demonstrated either [1,2] and/or [2,3] rearrangements, with the selectivity greatly dependent on the reaction conditions and substituents present (Scheme 3). Thus thermolysis of 4a in refluxing MeCN gave only the benzoxazepine $8a^7$, the product of Meisenheimer rearrangement. Heating 4b under the same conditions produced two inseparable products consistent with the benzoxazepine 8b and benzoxazonine 9b. Thermolysis in refluxing DCM, however, gave 9b (41%)⁸ selectively, most probably via a stereoselective [2,3] sigmatropic rearrangement of *cis*-4b, as *trans*-4b was recovered unchanged. The structure of 9b was confirmed only after low temperature NMR experiments⁸ with the broad peaks observed at 25°C resolved to two distinct forms, of equal population, at -30°C. The Z stereochemistry of 9b was confirmed by nOe experiments. Compound 9b isomerised almost completely to 8b (8b/9b = 19) upon refluxing in xylene for 1 h., indicating the less strained benzoxazepine ring is

thermodynamically favoured. Refluxing **9b** in MeCN for 1 h. produced only slight isomerisation ($\leq 7\%$) to **8b**.



Studies are continuing on factors affecting the competition between [2,3] and [1,2] rearrangements in appropriately substituted tetrahydroisoquinoline N-oxide and N-ylide derivatives and on the usefulness of these rearrangements in synthesis.

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- 5. All new compounds gave spectral data consistent with the proposed structures.

 Compound E-6a. ¹H NMR* δ: 6.77 (s, H8), 6.62 (s, H11), 6.43 (d, J 16.1 Hz, H7), 5.48 (m, H6), 4.17-4.04 (m, CH₂ of Et), 3.87 (s, OCH₃), 3.84 (s, OCH₃), 3.54 (dd, J 1.7 Hz, J 6.3 Hz, H4), 2.84-2.60 (m, 4H), 2.70 (s, NCH₃), 2.38-2.28 (m, 2H), 1.25 (t, CH₃ of Et); ¹³C NMR* δ: 173.25 (COO), 147.30 (C9^A), 147.09 (C10^A), 134.69 (C7), 133.87 (C7a^B), 132.09 (C11a^B), 130.64 (C6), 113.77 (C11), 110.66 (C8), 67.19 (C4), 60.58 (CH₂ of Et), 56.40 (2OCH₃), 56.00 (C2), 45.24 (NCH₃), 38.03 (C1), 36.32 (C5), 14.92 (CH₃ of Et); MS m/z: 319 (M⁺, 8%; Calcd. for C_{18H25}NO4 319.1783, found 319.178), 246 (100), 215 (7), 204 (7), 189 (12).

*NMR were run in CDCl₃ at 300MHz (¹H) and 75.5MHz (¹³C). Compound Z-**6a**. ¹H NMR δ : 6.67 (d, J 10.7 Hz, H7), 6.58 (s, ArH), 6.54 (s, ArH), 5.90-5.81 (m, H6), 4.10 (q, CH₂ of Et), 3.87 (s, OCH₃), 3.84 (s, OCH₃), 3.49-3.44 (m, H4), 3.14-3.07 (m, 1H2), 2.90-2.70 (m, 2H), 2.62-2.56 (m, 1H), 2.58 (s, NCH₃), 2.22-2.12 (m, 2H5), 1.23 (t, CH₃ of Et); ¹³C NMR δ : 173.60 (COO), 148.34 (C9^A), 147.53 (C10^A), 134.24 (C7a^B), 132.42 (C6^C), 131.60 (C7^C), 129.53 (C11a^B), 113.84 (C8^D), 111.89 (C11^D), 66.94 (C4), 60.88 (CH₂ of Et), 56.51 (2OCH₃), 53.07 (C2), 45.48 (NCH₃), 37.86 (C5), 32.63 (C1), 15.00 (CH₃ of Et); MS m/z: 319 (M⁺, 14%; Calcd. for C₁₈H₂₅NO₄ 319.1783, found 319.1778), 246 (100), 215 (2), 203 (3), 189 (6). Compound *E*-**6b**. ¹H NMR δ : 6.71 (s, ArH), 6.67 (s, ArH), 6.28 (s, H7), 4.12 (q, CH₂ of Et), 3.87 (s, OCH₃), 3.85 (s, OCH₃), 3.54 (d, J 6.3 Hz, H4), 2.88-2.70 (m, 3H), 2.65 (s, NCH₃), 2.59-2.53 (m, 1H), 1.45 (s, CH₃), 1.26 (t, CH₃ of Et); ¹³C NMR δ : 174.34 (COO), 147.29 (C9^A), 147.03 (C10^A), 133.87 (C7a^B), 132.48 (C11a^B), 131.64 (C6^B), 129.66 (C7), 113.84 (C8^C), 111.89 (C11^C).

(C10^A), 133.87 (C7a^B), 132.48 (C11a^B), 131.64 (C6^B), 129.66 (C7), 113.84 (C8^C), 111.89 (C11^C), 65.92 (C4), 60.82 (CH₂ of Et), 56.67 (OCH₃), 56.51 (OCH₃), 55.22 (C2), 45.64 (NCH₃), 41.56 (C5), 39.00 (C1), 18.91 (CH₃), 14.99 (CH₃ of Et); MS m/z: 333 (M⁺, 23%; Calcd. for C₁₉H₂₇NO₄ 333.1940, found 333.1922), 276 (10), 260 (100), 217 (5), 203 (10).

Compound Z-6b. ¹H NMR δ : 6.56 (s, ArH), 6.53 (s, ArH), 6.38 (s, H7), 4.09 (q, CH₂ of Et), 3.86 (s, OCH₃), 3.83 (s, OCH₃), 3.59 (dd, *J* 4.0 Hz, *J* 12.6 Hz, H4), 3.06-2.99 (m, 1H), 2.90-2.82 (m, 1H), 2.71-2.57 (m, 2H), 2.60 (s, NCH₃), 2.27 (dd, *J* 13.3 Hz, *J* 13.3 Hz, 1H5), 2.03 (dd, *J* 3.9 Hz, *J* 14.0 Hz, 1H5), 1.90 (s, CH₃), 1.23 (t, CH₃ of Et); ¹³C NMR δ : 174.31 (COO), 148.07 (C9^A), 147.20 (C10^A), 138.59 (C7a^B), 135.04 (C11a^B), 130.70 (C6^B), 127.53 (C7), 113.63 (C8^C), 112.10 (C11^C), 65.04 (C4), 60.65 (CH₂ of Et), 56.43 (2OCH₃), 52.95 (C2), 45.57 (NCH₃), 38.35 (C5), 36.65 (C1), 22.31 (CH₃), 15.01 (CH₃ of Et); MS m/z: 333 (M⁺, 21%; Calcd. for C₁₉H₂₇NO₄ 333.1940, found 333.1951), 260 (100), 229 (4), 203 (107), 130 (4).

- Compound 8a. ¹H NMR δ: 6.67 (s, ArH), 6.58 (s, ArH), 6.01-6.10 (m, H1'), 5.45 (d, J 6.8 Hz, H1), 5.31-5.22 (m, 2H2'), 3.86 (s, OCH₃), 3.83 (bs, OCH₃ and 1H), 3.28-24 (m, 1H), 3.10-3.02 (m, 1H), 2.92-2.85 (m, 1H), 2.72 (s, NCH₃); ¹³C NMR δ: 148.03 (C7^A), 147.89 (C8^A), 137.17 (C1'), 132.81 (C5a^B), 129.63 (C9a^B), 117.78 (C2'), 114.39 (C6^C), 111.16 (C9^C), 87.12 (C1), 60.69 (C4), 56.57 (2OCH₃), 47.39 (NCH₃), 33.25 (C5); MS m/z: 249 (M⁺, 22%; Calcd. for C₁₄H₁₉NO₃ 249.1363, found 249.135), 232 (10), 203 (11), 190 (100), 175 (44), 159 (22), 147 (17).
- Compound 9b. ¹H NMR (25°C) δ: 6.67 (s, ArH), 6.52 (s, ArH), 6.46 (s, H7), 4.06 (bs, 2H5), 3.86 (s, OCH₃), 3.82 (s, OCH₃), 2.88 (bs, 2H), 2.57 (s, NCH₃), 1.90 (s, CH₃), 2H of CH₂CH₂ not detected; ¹³C NMR (25°C) δ: 148.25 (C9^A), 147.13 (C10^A), 138.68 (C7a^B), 133.58 (C11a^B), 131.10 (C6), 127.93 (bs, C7), 111.58 (C8^C), 111.16 (C11^C), 74.77 (bs, C5), 62.81 (C2), 56.28 (OCH₃), 56.20 (OCH₃), 47.41 (bs, NCH₃), 33.24 (bs, C1), 24.36 (CH₃); MS m/z: 263 (M⁺, 12%; Calcd. for C₁₅H₂₁NO₃ 263.1521, found 263.1532), 246 (5), 204 (100), 189 (37), 175 (11).
 ¹H NMR (-30°C) doubling of most peaks δ: 6.78, 6.65, 6.56 and 6.54 (s, 2ArH), 6.49 (s, H7), 4.25-3.95 (4 doublets, 2 superimposed, from 2H5), 3.92 and 3.89 (s, OCH₃), 3.85 (s, OCH₃), 3.2-2.9 (m, 2H), 2.9-2.7 (m, 1H), 2.7-2.5 (m, 1H, obscured by NMe), 2.67 and 2.57 (s, NCH₃), 1.98 and 1.87 (s, CH₃); ¹³C NMR (-30°C) doubling of most peaks δ: 147.95, 146.98, 146.59 and 146.13 (C9 and C10). 139.25 and 137.30 (C7a^B), 133.65 and 132.08 (C11a^B), 130.48 (C6), 129.37 and 126.18 (C7),

110.66 and 110.49 (C8^C), 110.11 and 109.85 (C11^C), 77.25 and 72.46 (C5), 62.80 and 62.20 (C2),

55.99 (20CH₃), 48.64 and 46.29 (NCH₃), 34.44 and 31.35 (C1), 24.64 and 24.24 (CH₃).

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